

## TRITERPENOIDS FROM *GLOCHIDION HEYNEANUM*\*

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**Key Word Index** *Glochidion heyneanum*, Euphorbiaceae, triterpenoid glycosides, glochidioside N and Q

**Abstract**—Two new triterpene glycosides, glochidioside N and glochidioside Q were isolated from *G. heyneanum*, besides nine known compounds, glochidone, glochidonol, glochidiol, lup-20(29)-en-1 $\beta$ ,3 $\beta$ -diol, epimachaerinic acid,  $\beta$ -amyrin, stigmasterol, sitosterol- $\beta$ -D-glucoside and D-mannitol.

### INTRODUCTION

Earlier, we reported the isolation of glochidioside, a new triterpene glycoside, from *Glochidion heyneanum* [1]. In the present communication the isolation and structure elucidation of two new triterpenoid glycosides, glochidioside N and Q are reported along with a number of known triterpenes from this plant.

### RESULTS AND DISCUSSION

Repeated column chromatography of the *n*-butanol fraction yielded pure glochidioside N as an amorphous colourless solid. A positive Fiegel test revealed the presence of a carbohydrate moiety in glochidioside N. The IR spectrum displayed bands for a hydroxyl group (3400, 1100, 1070  $\text{cm}^{-1}$ ) and an aromatic ester group (1710, 1700, 1690, 1280  $\text{cm}^{-1}$ ). The presence of an aromatic ester group was also substantiated by UV absorption at 230 and 273 nm.

Substance N was subjected to acid hydrolysis. In the hydrolysate the sugar moiety was identified as glucose by paper chromatography. The aglycone portion was purified by column chromatography and crystallized from  $\text{CHCl}_3$ -MeOH into colourless crystals (1), mp 282–284°,  $[\alpha]_D^{+59}$  (EtOH; *c* 1). Compound 1 had the molecular formula  $\text{C}_{37}\text{H}_{54}\text{O}_6$  (FDMS,  $[\text{M}]^+$  at  $m/z$  594). The aglycone was finally characterized as 16-*O*-benzoyl-gymnestrogenin on the basis of mixed mp, IR, NMR and MS [1].

The position of the sugar linkage to the aglycone was fixed on the basis of the  $^1\text{H}$  NMR spectral analyses of the aglycone and peracetylated glochidioside N. In the  $^1\text{H}$  NMR spectrum of the peracetylated glochidioside N, the H-3 carbinolic proton resonated at the same frequency ( $\delta$  3.56) as that of the aglycone while the other carbinolic proton signals were shifted downfield due to acetylation. Thus C-3 was established as the position of the sugar linkage to the aglycone. The  $^1\text{H}$  NMR spectrum of the peracetylated glochidioside N had a  $^1\text{H}$  doublet at  $\delta$  4.66 ( $J = 8$  Hz) for the anomeric proton of the D-glucopyranose. The high magnitude of the  $J$  value suggested

that the sugar had a  $\beta$ -linkage and was in  $4C_1$  conformation. Thus, the complete structure of glochidioside N was established as 3 $\beta$ -[(*O*- $\beta$ -D-glucopyranosyl)oxy]-16 $\beta$ -benzoyloxy-olean-12-ene-21 $\beta$ ,23,28-triol [2].

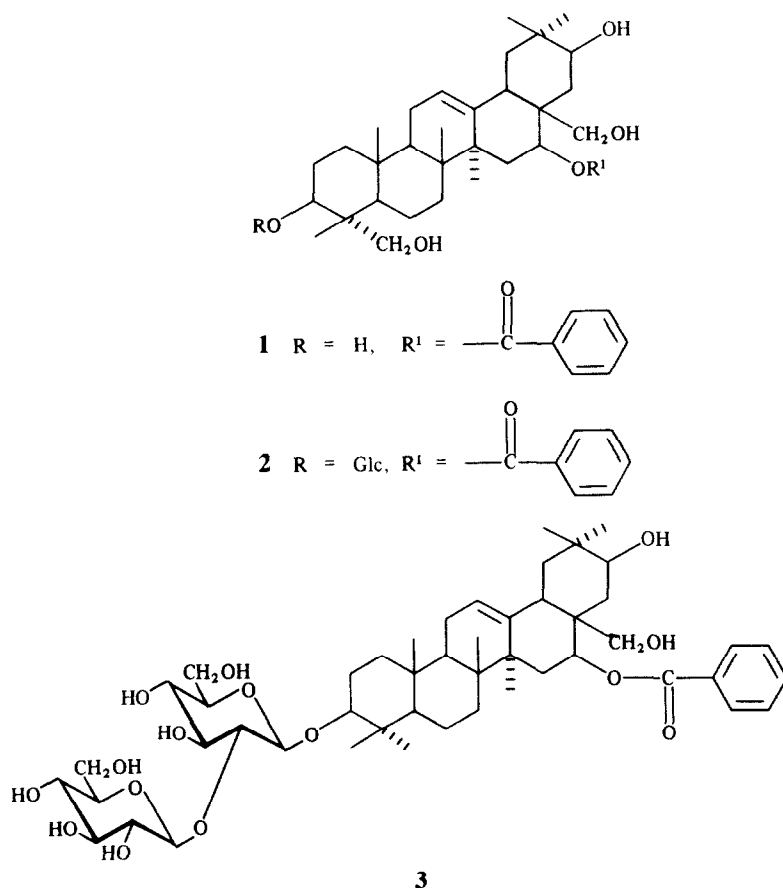
Glochidioside Q was also obtained as an amorphous solid which gave a positive Fiegel test revealing its glycosidic nature. The IR spectrum displayed bands at 3400, 1130 and 1080  $\text{cm}^{-1}$  for a hydroxy group and 1705, 1650 ( $\text{C}=\text{O}$ ) and 1280  $\text{cm}^{-1}$  for an aromatic ester group. Acid hydrolysis of substance Q furnished an aglycone and a mixture of sugars.

Chromatography of the aglycone portion yielded colourless needles, mp 282°,  $[\alpha]_D^{+59}$  (EtOH). The aglycone was found to be the same as obtained from glochidioside N, viz 16-*O*-benzoyl gymnestrogenin (1) on the basis of mixed mp, IR, NMR and MS [1].

Paper chromatography of the sugar portion along with other sugars revealed the presence of glucose as the sugar moiety, which was finally characterized as D-glucose,  $[\alpha]_D^{+48}$  ( $\text{H}_2\text{O}$ ; *c* 1). Again the site of linkage of the sugar to the aglycone was found to be C-3 by comparison of  $^1\text{H}$  NMR spectra of aglycone 1 with that of peracetylated glochidioside. It was found that apart from H-3 all the other carbinolic protons were shifted downfield due to acetylation, while H-3 resonated at the same frequency ( $\delta$  3.52). Therefore, in glochidioside Q the sugar moiety was located at C-3 of the aglycone.

In order to determine the linkage and number of sugar units in glochidioside Q, it was subjected to permethylation by the Hakomori's method [2]. The permethylated glycoside was hydrolysed with acid and the resulting partially methylated sugars were identified as 1,5-di-*O*-acetyl-2,3,4,6-tetra-*O*-methyl glucose and 1,2,5-tri-*O*-acetyl-3,4,6-tri-*O*-methyl glucose by GC/MS analyses of their alditol acetates according to Jansson *et al.* [3]. Evidently in glochidioside Q, the terminal glucopyranosyl unit was linked to the other glucopyranosyl unit through 1 $\rightarrow$ 2 linkage. The latter was in glycosidic linkage with the aglycone at C-3. The  $^1\text{H}$  NMR spectrum of the peracetylated glochidioside Q had two  $^1\text{H}$  doublets at  $\delta$  4.59 ( $J = 8$  Hz) and 4.50 ( $J = 8$  Hz) for the anomeric protons of the two glucose units. The high magnitude of  $J_{1,2}$  values suggested that the sugars were in  $4C_1$  (D) conformation and were linked together through  $\beta$ -linkages. Thus the structure of glochidioside Q has been

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established as 3 $\beta$ -[(*O*- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 2)-*O*- $\beta$ -D-glucopyranosyl)oxy]-16 $\beta$ -benzoyloxy-olean-12-ene-21 $\beta$ ,23,28 triol (3)

The chloroform-soluble fraction of the alcoholic extract of the plant on chromatography over silica gel yielded  $\beta$ -amyrin, glochidone, stigmasterol, glochidonol, glochidiol, lup-20(29)-en-1 $\beta$ ,3 $\beta$ -diol and epimachaerinic acid, while repeated column chromatography of the *n*-BuOH fraction yielded sitosterol- $\beta$ -D-glucoside and D-mannitol

Literature survey revealed the presence of glochidone, glochidiol, glochidonol, and lup-20(29)-en-1 $\beta$ ,3 $\beta$ -diol in other species of genus *Glochidion* also. This observation is significant from the view point of chemotaxonomy

#### EXPERIMENTAL

Mps uncorr. IR spectra in KBr pellets. The  $^1\text{H}$  NMR spectra were recorded at 90 or 400 MHz. Mass spectra were recorded with a JEOL JMS-D-300 mass spectrometer. Gas liquid chromatography was conducted in a chemito 3800 gas chromatograph. CC employed silica gel impregnated with 4% boric acid. TLC were performed on silica gel G and Silica gel G impregnated with 2% boric acid using the following solvent systems: (1) EtOAc-2% aq.  $\text{Me}_2\text{CO}$  (3:2), (2) EtOAc- $\text{Me}_2\text{CO}$ - $\text{H}_2\text{O}$  (8:8:1), (3)  $\text{CHCl}_3$ -MeOH (47:3), (4)  $\text{C}_6\text{H}_6$ -MeOH (23:2), (5)  $\text{C}_6\text{H}_6$ -MeOH (24:1), (6)  $\text{C}_6\text{H}_6$ . The compounds were visualized by spraying with 1%  $\text{Ce}(\text{SO}_4)_2$  in 1M  $\text{H}_2\text{SO}_4$ . PC was carried out on Whatman No. 1 paper using *n*-BuOH satd with  $\text{H}_2\text{O}$  as the eluant.

**Isolation of glochidoside N.** Repeated chromatography of *n*-BuOH soluble portion over silica gel impregnated with 4% boric acid using EtOAc (satd with  $\text{H}_2\text{O}$ )-MeOH (45:8) as the solvent system yielded glochidoside N (50 mg) as an amorphous powder. IR  $\nu_{\text{max}}^{\text{KBr}} \text{ cm}^{-1}$ : 3400 (OH), 2950, 1710, 1700, 1690, 1280, 1100, 1070.

**Acid hydrolysis of glochidoside N.** Glochidoside N (35 mg) was refluxed with 2N HCl in 80% aq. EtOH for 3 hr. After addition of  $\text{H}_2\text{O}$  followed by evapn of EtOH the hydrolysate was refluxed for a further 1 hr and then extracted with  $\text{CHCl}_3$  (3  $\times$  1 ml). The aqueous acidic phase was neutralised with Amberlite IR 410 ( $\text{CO}_3^{2-}$ ) resin. The organic layer was successively washed with aq.  $\text{NaHCO}_3$  soln and  $\text{H}_2\text{O}$ , then evapd to give a residue (20 mg) containing the aglycone.

**Aglycone.** The crude aglycone (20 mg) was purified by CC using  $\text{CHCl}_3$ -MeOH (47:3) eluate and crystallized from the same solvent system into colourless needles, mp 282–284°.  $[\alpha]_D^{25} + 59^\circ$  (EtOH, *c* 1), IR  $\nu_{\text{max}}^{\text{KBr}} \text{ cm}^{-1}$ : 3450 (OH), 2950, 1700, 1650, 1320, 1280, 1070, 730, UV  $\lambda_{\text{max}}^{\text{MeOH}} \text{ nm}$ : 230, 273, FDMS  $m/z$ : 594  $[\text{M}]^+$ , EIMS  $m/z$ : 576  $[\text{M}-18]^+$ .  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ): 3.70 (2H, *m*, H-3, H-21), 5.00 (1H, *q*, *J* = 12.5 Hz, H-16).

**Isolation and identification of sugar.** The neutralized aq. fraction of the hydrolysate was evapd to a syrup (8 mg) which was chromatographed over silica gel (1.0 g) and the fractions monitored by TLC in solvent system (2). Pure sugar was obtained in EtOAc (satd with  $\text{H}_2\text{O}$ ) + 30%  $\text{Me}_2\text{CO}$ . This was identified as D-glucose,  $[\alpha]_D^{25} 48^\circ$  ( $\text{H}_2\text{O}$ , *c* 1) by paper chromatographic comparison with the authentic sample.

**Acetylation of glochidoside N.** Glochidoside N (10 mg) was refluxed at 100° with  $\text{Ac}_2\text{O}$ -pyridine (1 ml each) for 2 hr, then

evapd to dryness. The residue was dissolved in  $\text{CHCl}_3$  and washed with  $\text{H}_2\text{O}$  ( $2 \times 1$  ml), then evapd to dryness. The residue thus obtained was purified by CC over silica gel using *n*-hexane– $\text{Me}_2\text{CO}$  (9/1) as eluant. It did not show any OH bands in the IR spectrum.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  0.78, 0.88, 0.96, 0.99 (6H), 1.14 (3H, s, Me), 1.99, 2.01, 2.05, 2.09, 2.10, 2.22, 2.13, (3H each, s,  $7 \times \text{OAc}$ ), 3.56 (1H, q,  $J = 10$ , 5 Hz,  $3\alpha\text{-H}$ ), 4.66 (1H, d,  $J = 8$  Hz, anomeric H), 4.87 (1H, q,  $J = 12$ , 6 Hz,  $21\alpha\text{-H}$ ), 5.34 (1H, m, 12-H), 5.62 (1H, q,  $J = 12$ , 5 Hz,  $16\alpha\text{-H}$ ), 7.43 (2H, t,  $J = 8$  Hz), 7.55 (1H, t,  $J = 8$  Hz), 8.03 (2H, d,  $J = 8$  Hz).

**Glochidioside Q (3)** Repeated CC over silica gel impregnated with 4% boric acid using  $\text{EtOAc}$ –2% Aq  $\text{Me}_2\text{CO}$  (7/3) as eluant yielded glochidioside Q as an amorphous powder (80 mg), IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$  3400(OH), 1130, 1080, 1705, 1650 ( $\text{C}=\text{O}$ ), 1280.

**Acid hydrolysis of glochidioside Q** Glochidioside Q (35 mg) was refluxed with 2N HCl in 80% aq EtOH for 3 hr,  $\text{H}_2\text{O}$  (1.5 ml) was added, EtOH was evapd and the reaction mixture was refluxed again for 1 hr. The reaction mixture was worked-up in the same manner as in the case of glochidioside N.

Paper chromatography of the aqueous portion (15 mg) as well as its  $[\alpha]_{\text{D}} + 48^\circ$  ( $\text{H}_2\text{O}$ , c 1) at equilibrium revealed the presence of D-glucose as the sugar moiety while the aglycone, mp  $282^\circ$ ,  $[\alpha]_{\text{D}} + 59^\circ$  ( $\text{C}_6\text{H}_5\text{OH}$ , c 1) was found to be the same as obtained from glochidioside N on the basis of mmp, IR, NMR and MS.

**Permethylation of glochidioside Q** Glochidioside Q (30 mg) was dissolved in dry DMSO (20.0 ml) in a 25 ml flask sealed with a rubber septum.  $\text{N}_2$  was passed in the flask with the help of two injection needles and dimethyl sodium (2M, 5.0 ml) was added by means of a syringe. The reaction mixture was stirred for 30 min and left at room temp overnight. After cooling the reaction mixture, MeI (8 ml) was added and the resultant mixture stirred at room temperature for 1 hr. Excess MeI was evapd under vacuum, and the reaction mixture poured into  $\text{H}_2\text{O}$  (20 ml) and extracted with  $\text{CHCl}_3$  ( $4 \times 15$  ml). The combined  $\text{CHCl}_3$  phase was washed with  $\text{H}_2\text{O}$  ( $4 \times 10$  ml) and evapd to dryness. The permethylated product (21.0 mg) was chromatographed over silica gel (2 g) using *n*- $\text{C}_6\text{H}_{14}$ – $\text{Me}_2\text{CO}$  as the solvent system. The *n*- $\text{C}_6\text{H}_{14}$ – $\text{Me}_2\text{CO}$  (9/1) eluate contained pure permethylated product. Its IR spectrum was devoid of hydroxyl bonds.

**Preparation of alditol acetates** Permethylated glochidioside Q (10 mg) was subjected to hydrolysis with  $\text{HCO}_2\text{H}$  at  $100^\circ$  for 1 hr and with 2N  $\text{H}_2\text{SO}_4$  at  $100^\circ$  for 2 hr. The hydrolysate was then extracted with  $\text{CHCl}_3$  and the aq. acidic phase was neutral-

ized with Amberlite IR 410 ( $\text{CO}_3^{2-}$ ) resin. The aq. portion was concd to 2 ml,  $\text{NaBH}_4$  (50 mg) added and stirred at room temp. After neutralization with Amberlite IR-120 ( $\text{H}^+$ ) resin,  $\text{H}_3\text{BO}_3$  formed in the reaction was removed by repeated evapn with MeOH, followed by acetylation with  $\text{Ac}_2\text{O}$ –pyridine (1/1, 2 ml) at  $100^\circ$  for 1 hr. The resulting mixture of alditol acetates was analysed by GC/MS on a column of 3% OV-225 at  $170^\circ$ . The GC showed two peaks, A and B (ratio 1/1 R, 13.0 and 7.2 min).

**Alditol acetate A** EIMS  $m/z$  189, 179, 161, 145, 101, 99, 87 for 1,2,5-tri-*O*-acetyl-3,4,6-tri-*O*-methyl glucitol.

**Alditol acetate B** EIMS  $m/z$  205, 161, 145, 129, 117, 101 for 1,5-di-*O*-acetyl 2,3,4,6-tetra-*O*-methyl glucitol.

**Peracetylated glochidioside Q** Glochidioside Q (10 mg) was acetylated in with  $\text{Ac}_2\text{O}$ –pyridine (1/1, 2 ml) in the usual manner.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  0.78, 0.88, 0.96, 1.0, 1.0 (6H), 1.14 (3H each, s, Me), 1.96, 1.97, 1.99, 2.01, 2.05, 2.06, 2.09, 2.10 (3H, each s,  $8 \times \text{CO}_2\text{Me}$ ), 3.52 (1H, q,  $J = 10$ , 5 Hz,  $3\alpha\text{-H}$ ), 4.50 (1H, d,  $J = 8$  Hz, anomeric H), 4.59 (1H, d,  $J = 8$  Hz, anomeric H), 4.86 (1H, q,  $J = 12$ , 6 Hz,  $21\alpha\text{-H}$ ), 5.35 (1H, m, 12-H), 5.61 (1H, q,  $J = 12$ , 5 Hz,  $16\alpha\text{-H}$ ).

**Glochidone (4)** This was crystallized from  $\text{CHCl}_3$ –MeOH as colourless needles, mp  $163$ – $164^\circ$ , TLC in system (6),  $[\alpha]_{\text{D}} + 70.2^\circ$  ( $\text{CHCl}_3$ , c 1) [lit mp  $164$ – $165^\circ$ ,  $[\alpha]_{\text{D}} + 73.41^\circ$ ], UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm 228, IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$  1660 ( $\alpha,\beta$  unsaturated ketone), 885 ( $=\text{CH}_2$ ). It was characterized as 4 on the basis of IR,  $^1\text{H NMR}$  and MS [4],  $^1\text{H NMR}$ ; Table 1.

**Glochidonol (5)** This was obtained as an amorphous solid from the repeated CC of the  $\text{CHCl}_3$  soluble portion over silica gel. TLC in system (5),  $[\alpha]_{\text{D}} + 46^\circ$  ( $\text{CHCl}_3$ , c 1) lit.  $[\alpha]_{\text{D}} + 49.4^\circ$ , IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$  3400 (OH), 1700 ( $\text{C}=\text{O}$ ), 1630 ( $\alpha,\beta$  unsaturated ketone), 880 ( $=\text{CH}_2$ )  $\text{cm}^{-1}$ . Its structure was established as [5],  $^1\text{H NMR}$  Table 1.

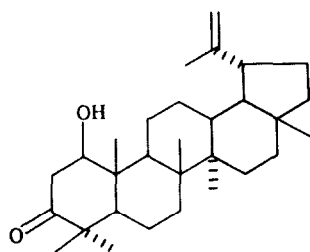
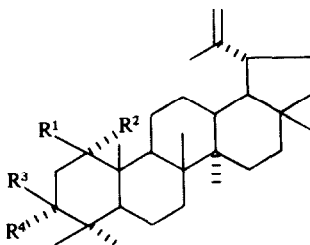
**Glochidiol (6)** This crystallized from MeOH– $\text{CHCl}_3$ , mp  $256^\circ$ . TLC in solvent (4),  $M^+$  at  $m/z$  442. Its IR, MS and  $^1\text{H NMR}$  data were in agreement with those of 6 [4–6].  $^1\text{H NMR}$  Table 1.

**Lup-20(29)-en-1 $\beta$ ,3 $\beta$ -diol (7)** It was crystallized from  $\text{CHCl}_3$ –MeOH, mp  $230$ – $232^\circ$  literature mp  $235^\circ$ . TLC in solvent system (4)  $M^+$  at  $m/z$  442. IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$  3360 (OH), 2970, 1630, 1450, 1380, 1050, 890 ( $=\text{CH}_2$ ). It was characterized by comparing the  $^1\text{H NMR}$  spectral data of its dihydrodiacetate with that reported in the lit [7].  $^1\text{H NMR}$  Table 1.

**Epimacherinic acid (8)** This was crystallized from EtOH as colourless needles, mp  $258$ – $260^\circ$ , IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$  3490 (OH), 2999

Table 1  $^1\text{H NMR}$  chemical shifts of compounds 4, 5, 6 and 7 in  $\text{CDCl}_3$  ( $J$  values are given in parentheses)

	4	5	6	7
Quaternary methyls	0.76, 0.90 (6H), 1.03, 1.06, 1.43, s	0.78, 0.84, 0.98 (6H), 1.04, 1.06, s	0.78, 0.82, 0.90 (6H), 0.97, 1.04, s	0.70, 0.73, 0.85 0.90 (6H), 0.99
Vinylc methyl	1.60, 3H, s	1.68, 3H, s	1.67, 3H, s	1.62
$-\text{C}=\text{CH}_2$	4.50, 1H, 1H, br s 4.60, 1H, br s	4.56, 1H, br s 4.69, 1H, br s	4.51, 1H, br s 4.61, 1H, br s	4.46, 1H, br s 4.56, 1H, br s
CHOH		3.91, 1H, q (8, 4)	3.48, 1H, m 3.78, 1H, q (12, 6)	3.18, 1H, q (12, 6) 3.35, 1H, q (10, 4)
$\text{CH}_2-\text{C}=\text{O}$	5.73, 1H, d, (10) 7.02, 1H, d, (10)	2.22, 1H, q (14, 4) 3.01, 1H, q (14, 8)		

**5**

**6** R<sup>1</sup> = OH, R<sup>2</sup> = H, R<sup>3</sup> = H, R<sup>4</sup> = OH

**7** R<sup>1</sup> = OH, R<sup>2</sup> = H, R<sup>3</sup> = OH, R<sup>4</sup> = H

(Me), 1700 (C=O), 1460, 1390, 1060, 833, <sup>1</sup>H NMR (90 MHz, CDCl<sub>3</sub> + DMSO-*d*<sub>6</sub> (2 drops) δ 0.70, 0.72, 0.82 (6H), 0.92 (6H), 1.06 (3H each, *s*, 7 × Me), 2.55 (1H, *d*, *J* = 2.5 Hz, H-18), 3.02 (1H, *q*, *J* = 12.6, CHOH), 3.35 (1H, *q*, *J* = 9.45 Hz, CHOH), 5.20 (1H, *m*, H-12) It was characterized as epimachaerinic acid [8] by

comparison with the data of the diacetate methyl ester reported in the literature

β-Amyrin, stigmasterol, sitosterol-β-D-glucoside and D-mannitol were identified on the basis of mmp, superimposable IR spectrum and co-TLC with the respective authentic samples

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